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Feng Xu, Wei-Xiao Hu,* Wei Zhou and Chun-Nian Xia

College of Pharmaceutical Science, Zhejiang University of Technology, Hangzhou 310032, People's Republic of China

Correspondence e-mail: huyang@mail.hz.zj.cn

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.009 Å R factor = 0.076 wR factor = 0.202 Data-to-parameter ratio = 7.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-Amino-2,5-bis(4-methoxybenzyl)-1,3,4-triazole

In the title compound, $C_{18}H_{20}N_4O_2$, the triazole ring is twisted with respect to the two benzene rings with dihedral angles of 71.5 (3) and 67.3 (3)°. N-H···N hydrogen bonding occurs between neighboring molecules.

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Comment

1,3,4-Triazole derivatives have extensive biological activity, including antibacterial and herbicidal action *via* effects on DNA and related molecules (Kahn & Martinez, 1998). Antiviral properties have also been found for some derivatives of 1-amino-1,3,4-triazole–ribofuranoside (Zakharieva *et al.*, 1994). We recently obtained the title triazole compound, (I), during the preparation of *s*-tetrazine derivatives.



The molecular structure of (I) is illustrated in Fig. 1. The molecule has a pseudo-twofold axis. The triazole ring plane is twisted with respect to the benzene rings, the dihedral angles being 71.5 (3) (C8-benzene ring) and 67.3 (3)° (C16-benzene ring). $N-H\cdots N$ hydrogen bonding occurs between neighboring molecules (Table 1).



Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

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Experimental

With sulfur (1.0 g) as a catalyst, 85% hydrazine hydrate (10 ml, 170 mmol) was dropped into *p*-methoxybenzyl cyanide (50 mmol) in anhydrous ethanol (15 ml) at 295 K. After refluxing for 3 h, the mixture was cooled to room temperature and the resulting solid product was filtered off. The solid product was dissolved in dichloromethane, affording single crystals of (I) by evaporation.

Z = 4

Crystal data

$C_{18}H_{20}N_4O_2$
$M_r = 324.38$
Monoclinic, Cc
a = 31.906 (16) Å
b = 6.130(3) Å
c = 8.726 (4) Å
$\beta = 98.428 \ (7)^{\circ}$
$V = 1688.3 (15) \text{ Å}^3$

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: none 3808 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.076$ $wR(F^2) = 0.202$ S = 0.961734 reflections 220 parameters H-atom parameters constrained $D_x = 1.276 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.09 \text{ mm}^{-1}$ T = 293 (2) K Prism, colorless $0.10 \times 0.08 \times 0.05 \text{ mm}$

1734 independent reflections 1100 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.078$ $\theta_{\text{max}} = 26.5^{\circ}$

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.1234P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.24 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.27 \text{ e} \text{ Å}^{-3}$ Extinction correction: *SHELXL97* Extinction coefficient: 0.021 (5)

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N6-H6A···N4 ⁱ	0.86	2.19	2.968 (7)	150
Symmetry code: (i) x.	v = 1, z			

Symmetry code: (1) x, y - 1, z.

Methyl H atoms were placed in calculated positions, with C–H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$. Other H atoms were placed in calculated positions, with C–H = 0.93 Å and N–H = 0.86 Å, and refined in riding mode, with $U_{iso}(H) = 1.2U_{eq}(C,N)$. In the absence of significant anomalous scattering effects, Friedel pairs were averaged.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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